## Photochemical Rearrangement of NN-Dimethylphenylethynylamine into $\alpha$ -Phenylisobutyronitrile

By R. Selvarajan and J. H. Boyer\*

(Chemistry Department, University of Illinois, Chicago Circle Campus, Chicago, Illinois 60680)

Summary Irradiation at 254 nm isomerized NN-dimethylphenylethynylamine (1) in cyclohexane into  $\alpha$ -phenylisobutyronitrile (3) by a new molecular rearrangement.

A solution of 1.0 g (7 mmol) of (1) in 250 ml of degassed anhydrous cyclohexane was irradiated in a Rayonet Reactor at 254 nm for 24 h. Monitoring by i.r. detected the appearance of a new band at 2020 cm<sup>-1</sup> (C:C:N)<sup>2</sup> which slightly increased as irradiation was continued an additional 96 h. Irradiation was stopped after an amorphous, apparently opaque, brown deposit covered the walls of the tube. Removal of cyclohexane left 0.927 g of a brown liquid: i.r. (CCl<sub>4</sub>) 2220 (vs doublet, C:C and C:N) and 2020 cm<sup>-1</sup> (s, C:C:N); n.m.r. (CDCl<sub>3</sub>)  $\delta$  1.69 [s, PhC  $(CN)(CH_3)_2$ ], 1.98 [s, PhC(CH<sub>3</sub>): C: NCH<sub>3</sub>], 2.80 [s, PhC- $CN(CH_3)_2$ , and 3.28 [s,  $PhC(CH_3):C:NCH_3$ ]. Of three volatile components detected by g.c. (5% poly-m-phenyl ether, 6 ft × 0.25 in, 120°) two were identified by comparison with authentic material as NN-dimethylphenylacetamide,<sup>3</sup> in trace amount, and α-phenylisobutyronitrile (3), 7.08 mg (2.43%). Unchanged (1) apparently polymerized during g.c. and was estimated by n.m.r. to be 0.639~g~(63%) with anisole as an internal standard. The product (3) (Equation 1) was identical with authentic material in comparison with g.c. retention time, i.r., and n.m.r.

Preliminary experiments have revealed no photo-sensitisation when irradiation at 300 nm was carried out in the presence of acetophenone, thioxanthen-9-one, naphthalene, butane-2,3-dione, 7H-benz[d,e]anthracen-7-one, or anthracene ( $E_T$  74, 65, 61, 55, 47, and 42 kcal mol<sup>-1</sup> respectively). When irradiated in 250 ml of anhydrous benzene at 254 nm for 120 h, 0.5 g of the ynamine (1) was isomerized into the nitrile (3) to the extent of 0.13%.

We acknowledge financial support from a NASA grant.

(Received, June 12th, 1970; Com. 901.)

<sup>1</sup>H. G. Viehe and M. Reinstein, Angew. Chem. Internat. Edn., 1964, 3, 506; Neth. Appl. 6,415,094, Chem. Abs., 1966, 64, 2006b.

<sup>2</sup> E. M. Burgess, R. Carithers, and L. McCullagh, J. Amer. Chem. Soc., 1968, 90, 1223 reported i.r. absorption at 2050 cm<sup>-1</sup> for PhCH; C: NPh.

<sup>4</sup> An authentic sample was prepared according to M. Haller and E. Bauer, Compt. rend., 1912, 155, 1582.

<sup>&</sup>lt;sup>3</sup> It is necessary to assume trace amounts of water in cyclohexane to account for hydration of the ynamine (1) into NN-dimethylphenylacetamide. H. G. Viehe, R. Fuks, and M. Reinstein, Angew. Chem. Internat. Edn., 1964, 3, 581 have observed a high sensitivity of ynamines toward moisture. Irradiation of the ynamine (1) in degassed spectrograde methylene chloride gave NN-dimethylphenylacetamide, m.p. 42—43° (Taverne, Rec. Trav. chim., 1897, 16, 37).